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volume than saturated solutions unsaturated solution is the solution which has a tendency of solute to get mixed in the solution up to its saturated point is called unsaturated solution the main example of unsaturated solution is sand in water types of saturation there are basically three types of saturation in any solution which can be discussed as follows 1 distinguish unsaturated saturated and supersaturated solutions the major component of the solution is called solvent and the minor component's are called solute if both components in a solution are 50 the term solute can be assigned to either component an unsaturated solution is a solution that contains less than the maximum amount of solute that is capable of being dissolved the figure below illustrates the above process and shows the distinction between unsaturated and saturated najam academy 703k subscribers 8.5k 363k views 3 years ago chemistry in this animated lecture i will teach you about saturated solution unsaturated 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of an unsaturated solution is a teaspoon of sugar in a glass of water if one adds a teaspoon of sugar to a glass of water it dissolves and one can still add more sugar to it because it is still unsaturated however at some point the sugar no longer dissolves in the water because the substance is saturated

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