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strong hydrogen bonds h bonds between water molecules exclude hydrocarbon chains from the body of the water accept hydrophilic part group forms hydrogen bonds h bonds to water accept hydrophobic section instead of hydrocarbon chain

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heterogeneous catalyst is in different phase than reactants and homogeneous catalyst in same phase homogeneous catalysts chemically change react and are reformed at end of reaction or reactants adsorb onto heterogeneous catalyst and products desorb

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m 25 00 g density of solution is 1 000 g cm 3 same as water or 25 00 cm3 solution has a mass of 25 00 g or mass of zinc reactant is negligible or mass of contents was 25 00 g c 4 18 j g 1 k 1 specific heat of solution is 4 18 j g 1 k 1 same as water

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3 c ii $14\ 950\ 392\ 19x$ answer must be given to $147\ 62\ x\ 37\ 74\ \mu g\ kg\ 1$ four significant figures do not accept values obtained directly from the graph $1\ 3\ c$ iii vanadium reduced in first reaction and oxidized in second reaction or v2o5 oxidizes so2 in first reaction and vo2 reduces o2 in second reaction or

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compare the ease of oxidation of s block and d block metals to their melting points and densities use section 25 of the data booklet 2

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by extrapolation of the two best fit lines required to give the highest temperature or

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once a correct answer to a question or part question is seen ignore further working even if this working is incorrect and or suggests a misunderstanding of the question this will encourage a uniform approach to marking with less examiner discretion

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instructions to candidates this paper consists of three 3 sections a b and c answer all questions in section a and b choose one 1 question from section c write your examination number on each page write all answers in the space provided use a blue or black pen in writing

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