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phase or gas phase in the case of liquid phase solutions the solvent is typically the compound question 1 state whether the following statements are correct or incorrect correct those which are wrong i when two positive integers are added we get a positive integer solution correct five examples a 46 83 129 for example 0 25 mol l 1 or 0 25 m solution of naoh means that 0 25 mol of naoh has been dissolved in one litre of one cubic

decimetre example 2 2example 2 2example 2 2calculate the molarity of a solution containing 5 g of naoh in 450 ml solution moles of naoh 1 5 g 40 g mol 0 125 mol volume of the solution in litres 450 ml

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for example 0.25 mol l<sup>-1</sup> or 0.25 m solution of naoh means that 0.25 mol of naoh has been dissolved in one litre or one cubic decimetre example 2 2example 2 2example 2 2calculate the molarity of a solution containing 5 g of naoh in 450 ml solution moles of naoh 1.5 g 40 g mol<sup>-1</sup> 0.125 mol volume of the solution in litres 450 ml

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