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introduction 1 1 purpose to outline the procedure for operation of typical ph meter in the traces centre and the undergraduate laboratories this procedure describes how to accurately measure the ph of a solution 1 2 scope applicable to digital ph meters located in traces centre and ug laboratory

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pen type digital ph meter all in one ph electrode is included easy to use data h old function for freezing the desired value water proof and protection auto power off without any operations after 10 mins supplied with a set of ph 4 01 ph 6 86 ph 9 18 buffers microcomputer circuit intelligent function high accuracy

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regularly calibrating your ph meter adjusts the electrode reading based off any changes and ensures that your readings are accurate and repeatable

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a ph meter is a piece of laboratory equipment used to measure hydrogen ion concentrations i e acidity and alkalinity in a water based solution expressed as ph it measures the difference in electrical potential between the ph electrode and the reference electrode

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a ph meter is a statistical tool that monitors the hydrogen ion activity in water based solutions determining its acidity or alkalinity represented as ph

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a three point calibration is required for the exstik ph meter the standard buffers used are ph 4 ph 7 and ph 10 dispense a small amount of the required buffers into a smaller container or tube for calibration use ph buffers for calibration one time only do not reuse

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