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of the known substance use the periodic table for grams 22 414 l of a gas at
stp or 6.022×10^{23} atoms or molecules 3 apply theories and rules of
chemistry to solve problems assess areas of strength and weakness for review
purposes improve problem solving strategy and learning efficiency
stoichiometry calculation practice worksheet 1 calculate the number of moles
of naoh that are needed to react with 500.0 g of h₂so₄ according to the
following equation h₂so₄ + 2 naoh → na₂so₄ + 2 h₂o ans 10.19 mol stoichiometry
quantitative relationships among substances as they participate in chemical

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stoichiometry quantitative relationships among substances as they participate in chemical

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