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equilibrium is said to be attained in reversible reaction when a all
the reactants have been used up b all the products have been
formed c there is no further change in temperature d the rates of
the forward and backward reactions are equal

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measures 3 level of awareness of the inter relationship between
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chemistry 2022 waec past questions a i define the first ionization energy of an element ii consider the following table and use it to answer the question that follows explain briefly why the first ionization energy of b is less than that of be despite the fact that the atomic number of b is greater than that of be

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