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gas one of the three fundamental states of matter with distinctly different properties from the liquid and solid states the remarkable feature of gases is that they appear to have no structure at all they have neither a definite size nor shape whereas ordinary solids have both a definite size a pure gas may be made up of individual atoms e g a noble gas like neon elemental molecules made from one type of atom e g oxygen or compound molecules made from a variety of atoms e g carbon dioxide a gas mixture such as air contains a variety of pure gases a gas is defined as a state of matter consisting of particles that have neither a defined volume nor defined shape it is one of the four fundamental states of matter along with solids liquids and plasma under ordinary conditions the gas state is between the liquid and plasma states a gas is a state of matter that has no fixed volume or shape in other words a gas takes the shape and volume of its container gases condense into liquids or ionize into plasma the particles in a gas are more distant from each other than in liquids gas particles may be elements or compounds gas laws overview dalton s law law of partial pressures the ideal gas law the volume of a given gas sample is directly proportional to its absolute temperature at constant pressure charles s law the volume of a given amount of gas is inversely proportional to its pressure when temperature is held constant boyle s law to describe the characteristics of a gas the three common phases or states of matter are gases liquids and solids gases have the lowest density of the three are highly compressible and completely fill any container in which they are placed states of matter gas the simplest of all states of matter what is gas gas is the state of matter in which the particles are far apart fast moving and not organised in any particular way gases are substances that exist in the gaseous state which is one of the three fundamental states of matter article vocabulary 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