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there are essentially two theories of solutions that can be considered exact the mcmillan mayer theory and fluctuation solution theory fst the first is mostly limited to solutes at low concentrations while fst has no such issue it is an exact theory that can be applied to any stable solution regardless of the number of components and their concentrations and the types of molecules and their sizes fluctuation theory of solutions applications in chemistry chemical engineering and biophysics outlines the general concepts and theoretical basis of fst and provides a range of applications described by experts in chemistry chemical engineering and biophysics the book which begins with a historical perspective and an introductory chapter includes a basic derivation for more casual readers it is then devoted to providing new and very recent applications of fst the first application chapters focus on simple model binary and ternary systems using fst to explain their thermodynamic properties and the concept of preferential solvation later chapters illustrate the use of fst to develop more accurate potential functions for simulation describe new approaches to elucidate microheterogeneities in solutions and present an overview of solvation in new and model systems including those under critical conditions expert contributors also discuss the use of fst to model solute solubility in a variety of systems the final chapters present a series of biological applications that illustrate the use of fst to study cosolvent effects on proteins and their implications for protein folding with the application of fst to study biological systems now well established and given the continuing developments in computer hardware and software increasing the range of potential applications fst provides a rigorous and useful approach for understanding a wide array of solution properties this book outlines those approaches and their advantages across a range of disciplines elucidating this robust practical theory

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adsorption from solution discusses the significance of adsorption behavior in thermodynamic terms with emphasis on the interplay between enthalpic and entropic contributions to the free energy this book examines the role of simple models and of elementary thermodynamic and statistical mechanical arguments in relation to the concept of surface phase organized into 22 chapters this book starts with an overview of the theoretical model for the solid liquid interface this text then proceeds with a discussion of the general thermodynamic treatment of adsorption from mixed solvents which is designed to apply in situations where adsorbed species may be regarded as distinct from their bulk counterparts other chapters discuss the adsorption from solutions of various interfaces of liquid gas liquid liquid or liquid solid the final chapter deals with the roles of adsorption from solution in controlling other phenomena such as liquid liquid displacement wetting and the forces between colloidal particles physicists chemists and materials scientists will find this book extremely useful

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